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ous systems involving salt-type minerals are examined in this work. Stability relations have been discussed on the basis of bulk equilibria.

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MINERAL-SOLUTION EQUILIBRIA IN SPARINGLY SOLUBLE

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Garrels, R.M. and Christ, C.L. (1965) Solutions, Minerals ...

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OCLC Number: 21346795: Notes: "A second edition based on the earlier book, Mineral equilibria, written by Dr. Garrels and published in 1960, in the Harper's geoscience series"--Page vii.

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SOLUTION-MINERAL EQUILIBRIA PART 1: CARBONATES 67 Thus the solubility is raised to a figure about twice as large as the 6.8×10^{-5} obtained by simply taking the square root of the solubility product [Eq. (3-7)]. At the same time the OH⁻ concentration is lowered slightly, so that the pH is approximately 9.9 instead of 10.0.

SOLUTION-MINERAL EQUILIBRIA PART 1: CARBONATES

Solutions, Minerals, and Equilibria Robert Minard Garrels, Charles Louis Christ Snippet view - 1965. Common terms and phrases. activity coefficients approximately aqueous association assumed atmosphere total pressure becomes behavior boundary cal/deg calculated carbonate cations Chapter chemical complex composition compounds concentration ...

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revision of R. M. Garrels, Mineral Equilibria with additional material by C. L. Christ. It is a substantial book of 450 pages, including a useful appendix of thermochemical constants of ions (in aqueous solution), and crystalline

BOOK REVIEWS

The book *Solutions, Minerals, and Equilibria* co-authored in 1965 by Garrels and Charles L. Christ revolutionized aqueous geochemistry. Garrels earned a bachelor's degree in geology from The University of Michigan in 1937. He went on to earn an M.S. degree from Northwestern in 1939, his thesis work was on iron ores of Newfoundland in 1938.

Robert Garrels - Wikipedia

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