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2.8 An antifreeze solution is prepared from 222.6 g of ethylene glycol, (C₂H₆O₂) and 200 g of water. Calculate the molality of the solution. Calculate the molality of the solution. If the density of the solution is 1.072 g mL⁻¹, then what shall be the molarity of the solution?

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$A = 4.2 \text{ kg}$; $m_B = 1.8 \text{ kg}$; $\theta = 32^\circ$; $a = 5.302 \text{ m/s}^2$ Required: tension in the string, F_T Analysis: We can substitute the value of acceleration into either of the equations from part (a) to solve for F_T . We will use Equation (1) because it is a bit simpler. Solution: $m_A g - F_T = m_A a$ $F_T = m_A (g - a)$ $F_T = (4.2 \text{ kg})(9.8 \text{ m/s}^2 - 5.302 \text{ m/s}^2)$...

Section 2.3: Applying Newton's Laws of Motion Tutorial 1 ...

Tutorial 2 Practice, page 20 1. Given: $v_i = 22 \text{ m/s}$ [up]; $v_f = 0 \text{ m/s}$; $a = 9.8 \text{ m/s}^2$ [down] Required: maximum height, d Analysis: Use $v_f^2 = v_i^2 + 2ad$ to calculate d . $0 = (22 \text{ m/s})^2 + 2(-9.8 \text{ m/s}^2)d$ $d = \frac{-(22 \text{ m/s})^2}{2(-9.8 \text{ m/s}^2)}$ $d = 25 \text{ m}$ Statement: The maximum height ...

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